

MAROS, Dezideriu, ing.; CSULAK, Acatiu, ing.; LEWY, Ladislau, ing.

Profiling the worm cutters for the machining of ratchet wheels by rolling method. Metalurgia constr mas 13 no.10:880-884 0 '61.

(Metal-cutting tools)
(Rolling(Metalwork))

CSULAK, A.

A method of geometric calculation of corrected cylindrical gearings.
Bul stiint polit Cluj 6:279-293 '63.

CSULAK, A.; SZEKELY, I.

Contributions to the multiple gearing theory of spur
gear wheels. Bul stiint polit Cluj no.7:255-262 '64.

CSULLAG, Jozsef, dr.

Eosinophilic granuloma of the skull-cap, appearing in the form of multiple myeloma. Magy. radiol. 15 no.5:299-302 S '63.

1. A Pécsi Orvostudományi Egyetem I sz. Belklinika (igazgató: Barta Imre dr. egyetemi tanár) közleménye.

(SKULL NEOPLASMS) (EOSINOPHILIC GRANULOMA)
(MULTIPLE MYELOMA) (DIAGNOSIS, DIFFERENTIAL)
(BIOPSY) (RADIOGRAPHY) (NEOPLASM RADIOTHERAPY)

LAMPE, Istvan; H. TOMITS, Gabriella; CSULLOG, Ferenc

Our audiometric and otoneurological observations on glaucoma patients. Szemeszet 100 no.3:145-149 S '63.

1. Debreceni Orvostudományi Egyetem Ful-orr-gegeklínikájának
(Igazgató: Jakabfi Imre egyet. tanár) és Szemklínikájának
(Igazgató: Kettesy Aladar egyet. tanár) közleménye.
(GLAUCOMA) (NYSTAGMUS) (COCHLEA)
(AUDIOMETRY) (VESTIBULAR FUNCTION TESTS)

CZECHOSLOVAKIA

CSUPKA, S; PETRASOVA, M; CARACH, J.

Section on Radiation Hygiene (Oddelenie radiačnej hygieny),
KHES, Bratislava - (for all).

Prague, Ceskoslovenska hygiena, No 10, December 1965, pp 615-617

"Content of Sr⁹⁰ and Cs¹³⁷ in radioactive deposit 1964."

APPROVED FOR RELEASE: Thursday, July 27, 2000

Card 2/2

L 24154-66 EWA(h)

ACC NR: AP6011979

SOURCE CODE: CZ/0038/66/000/001/0016/0019

AUTHOR: ~~Csupka, Stefan~~—~~Chupka, Sh.~~; ~~Petrasova, Maria~~—~~Petrashova, M.~~; ~~Carach, Jozef~~—~~Tsarakh, I.~~

ORG: Regional Hygiene-Epidemiological Station, Bratislava (Krajska hygienicko-epidemiologicka stanica)

TITLE: Contamination of the biosphere by sup 137 Cs from weapon test fallout

SOURCE: Jaderna energie, no. 1, 1966, 16-19

TOPIC TAGS: cesium radioactive fallout, atmospheric contamination, radioactive contamination

ABSTRACT: The concentration of ^{137}Cs in the precipitation on the territory of western Slovakia was measured in 1962 and 1963. The total β -active precipitation had, in 1963, a tendency to decrease. The specific gravity of the semi-monthly activity of ^{137}Cs in the total β activity amounted in 1962 to 0.9% and in 1963 to 3.2%. In 1963 the activity of ^{137}Cs was 1.3 times higher than the ^{90}Sr . The cumulative value of the radiocesium in 1962 to 1963 amounted to 37.8 nC/m^2 . This paper was presented by F. Behounek. The authors thank Academician F. Behounek for critical comments and useful advice. Orig. art. has: 2 figures and 2 tables. NA

SUB CODE: 18 / SUBM DATE: none / ORIG REF: 005 / OTH REF: 010 / SOV REF: 001

Card 1/1

UDC: 546.36.02: 621.039: 614.7(437)

CZECHOSLOVAKIA

CSUPKA, Stefan; CARACH, Jozef; PETRASOVA, Maria; Krajska Station of Hygiene and Epidemiology (Hygienicko Epidemiologicka Stanice), Bratislava.

"Content of Radioactive Strontium and Caesium in the Roots and Leaves of Vegetables in 1964, and the Ecology of the Plants."

Bratislava, Biologia, Vol 21, No 6, 1966, 432 -437

Abstract: The content of Sr^{90} and Cs^{137} in vegetables in Western Slovakia was investigated during 1964. During the spring season the vegetables contained nearly twice as much of the radioactive elements as during the fall. The leaves contained about $1\frac{1}{2}$ times as much of the radionuclides as the roots. A daily intake of Sr^{90} per inhabitant based on the consumption of vegetables averaged 2.8 pCi of Sr, and 4.2 pCi of Cs^{137} . 4 Tables, 7 Western, 2 Czech, 1 Russian reference. (Manuscript received 10 Nov 65).

1/1

PETRASOVA, M.; CSUPKA, S.; CARACH, J.

The results of radioactivity measurements of dust and rain falls in western Slovakia in 1961-1963. Cesk. hyg. 9 no.10:595-600
D ' 64.

1. Oddelenie radiacnej hygieny Krajskej hygienicko-epidemiologickej stanicy, Bratislava.

CSUPKA, S.; PETRASOVA, M.; CARACH, J.

Content of Sr90 and Cs137 in radioactive deposit in 1964.
Cesk. hyg. 10 no.10:615-617 D '65.

1. Oddelenie radiacnej hygieny Krajskej hygienicko-epidemiologickej stanice, Bratislava.

CSUPOR, L.

"Economy of Materials in Some of our Machine Factories", P. 5,
(TOBBTERMELES, Vol. 8, No. 8, July 1954, Budapest, Hungary)

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4,
No. 1, Jan. 1955, Uncl.

LIMPAR, Istvan, okleveles kohomernok; CSURBAKOVA, Tatjana, okleveles
kohomernok

Bar formation tests on deep-drawn aluminum sheets. Kch
lap 97 no.10:470-475 0 '64.

CSURES, Z.

Scientific results and tasks in light industry; also, remarks by A. Kiss and others. p. 75 KOZLEMENYEI. Budapest. (Reports issued by the Section of Technical Sciences, Hungarian Academy of Sciences. Quarterly) Vol. 14, No. 1/3 1954

SOURCE: East European Accessions List (EEAL) Library Of Congress
Vol. 5, No. 6, June 1956

CSURGAI, Imre

Making preparations for the repair of the one hundred thousandth car. Magy vasut 7 no.2:3 17 Ja '63.

1. Mühelybizottsági titkar.

CSURGAI, Lajos; KORTVELYES, Istvan

Chemicalization in agriculture. Elet tud 17 no. 16:503-506
22 Ap '62.

CSURGAY, Arpad

Signal transmission problems of moving points. Hir techn
ll no.4:151-159 Ag '60.

1. Tavkozlesi Kutato Interzet.

CSURGAY, Arpad

Designing inhomogeneous transmission line sections with prescribed reflection. Magyar techn 12 no.3:85-93 Je '61.

1. Tavkozlesi Kutato Intezet.

85632

H/009/60/000/004/001/001
A211/A026

6.2000 (2903, 3003, 3203, 3303)

AUTHOR: Csurgay, Árpád

TITLE: On Signal Transmission⁶ Problems of Moving Points

PERIODICAL: Híradástechnika, 1960, No. 4, pp. 151 - 159

TEXT: In the investigation of signal transmission of moving objects it was supposed that transmitter and receiver, which are to be connected by the signal channel, are in inertial position or, at least, that their speed could be ignored since their speed compared with the speed of light is negligible. The signal transmission of rockets traveling at light speed or coming close to it presents a different problem and can be solved by the general examination of signal transmission problems of moving objects applying the special theory of relativity. To determine the quality of intelligence transmitted between two such moving objects, the author deals with two problems. First he describes the electromagnetic field of antennas moving with high velocity, and secondly he determines the amplitude and frequency radiation pattern of transmitter and receiver antenna.¹⁵⁸ He arrives by various equations at a final formula which represents the type of the intelligence transmitted as a time function:

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On Signal Transmission Problems of Moving Points

$$f_B(t) = \frac{A(\varphi', \nu')}{r_0} \chi(1 + \frac{v}{c} \cos \varphi') \sum_{i=1}^{\infty} C_i e^{j\omega' i} \chi(1 + \frac{v}{c} \cos \varphi') (t - \frac{nr_0}{u}). \quad (49)$$

This equation contains 1) the transformed equation representing frequency radiation pattern $\omega = \omega' \chi(1 + \frac{v}{c} \cos \varphi')$ (39), where ω is the frequency received,

ω' the frequency transmitted and φ' the angle formed by the antenna and the direction in which it moves; 2) the transformed equation representing the amplitude radiation pattern $E = \frac{A}{r'} \chi(1 + \frac{v}{c} \cos \varphi')$ (41), where A is the transmitting antenna and r' the distance between antennas. The last section of the equation (49) is the equation $f_A(t') = \sum_{i=1}^{\infty} C_i e^{j\omega' i t'}$ (47), representing trans-

mitted $f_A(t')$ function given in the Fourier-series as sum of its harmonic components. When the radiation pattern of the antenna is $A(\varphi', \delta')$, the discernible field intensity is

$$E' = \frac{A(\varphi', \nu')}{r'} \sum_{i=1}^{\infty} C_i e^{j\omega' i} (t - \frac{r' r^i}{r} \cdot \frac{1}{c}), \quad (48)$$

which is in fact a flat wave propagating in the $\frac{r'}{r}$ direction. The author inves-

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investigates the distortion of electromagnetic waves at some specific distances between transmitter and receiver with the help of the equation (49). It is apparent from this equation that amplitude as well as frequency of different components is variable in relation to the function of time. Through a comparison of the Fourier series (48) and (49) it can be established that the ratio of the individual amplitude and frequency components is constant at any moment and equal to the ratio of the original values of the corresponding components. The absolute band width and the carrier frequency change proportionately with the time. The signal sent by the transmitter is expressed by the equation

$$f_A(t') = \frac{\tau'}{\pi} \int_0^{\frac{\omega \tau'}{2}} \frac{\sin \frac{\omega' \tau'}{2}}{\frac{\omega' \tau'}{2}} \cos \omega' t' \cdot d\omega' \quad (62)$$

and the received signal by the equation

$$f_B(t) = A_0 g_2 \frac{\tau'}{g_1} \int_0^{\frac{\omega \tau'}{2}} \frac{\sin \frac{\omega' \tau'}{2}}{\frac{\omega' \tau'}{2}} \cos \omega(t - \frac{nr_0}{c}) d\omega, \quad (64)$$

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where g_1 is the frequency change; g_2 the amplitude change in time function, and τ the time of impulses sent by the transmitter. The author further shows by calculations how the signals would change through the relative velocity between transmitter and receiver in four hypothetical transmissions. In the first example the author assumes that a rocket, traveling at a distance of 1,740,000 km from earth at a speed half that of light, would send SOS signals to a station on earth. The signals received on earth would be as presented in Figure 10. In the second example, the author assumes that two rockets are traveling away from the earth, one at a speed of $0.385x$ and the other at $0.6x$. Were a station on earth to transmit the Beethoven's Fifth Symphony to these rockets, then the first would receive this transmission one quint lower and each tone would be 1.5 times longer than the original one. The other would receive this transmission one octave lower and each tone would be two times longer. In the third example, the author assumes that a rocket approaching the earth were to send a one-hour comment to earth. How this transmission would be received on earth is shown in the equation

$$\frac{\tau}{\tau} = \frac{1 - \frac{v}{c}}{\sqrt{1 - (\frac{v}{c})^2}} = \sqrt{\frac{1 - \frac{v}{c}}{1 + \frac{v}{c}}}$$

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On Signal Transmission Problems of Moving Points

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The last example assumes a journey to Sirius. To the traveler this journey would last 12 years, but on earth 15 years would elapse. A continuous 12-year comment transmitted during this journey from the rocket to the earth would be heard over a period of 24 years on earth. A return journey from Sirius would appear also as a 12-year journey to the traveller but on earth again 15 years would elapse, whereas the transmission of a comment during the return journey would be heard on earth only over a period of 6 years. There are 11 figures and 5 references: 3 Hungarian and 2 German.

ASSOCIATION: Távközlési Kutató Intézet (Telecommunication Research Institute)

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H/009/61/000/003/001/001
D020/D105

9,1400

AUTHOR: Csurgay, Árpád

TITLE: Designing lengths of inhomogeneous transmission lines with prescribed reflection

PERIODICAL: Magyar Híradástechnika, no. 3, 1961, 85-93

TEXT: In investigating the problem of a length of an inhomogeneous transmission line, used for matching junctions of two transmission lines with different characteristic impedance, the author deals with (1) determining of the reflection function of a transition with an arbitrary contour and (2) determining of the geometrical configuration of a transition defined by a given reflection function. In the first section of his work, the author defines the four-pole parameters of an inhomogeneous waveguide, based on an article by G. Reiter (Ref. 10: Generalised Telegraphist's Equation for Wave Guides of Varying Cross-section, Convention on Long-Distance Transmission by Waveguide, London, Jan., 1959, Proc. IEE, Part B.) and Willis - Sinha (Ref. 8: Nonuniform Transmission Lines as

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Designing lengths of inhomogeneous transmission lines with prescribed reflection

Impedance Transformers, Proc. IEE, Part B, March, 1956), and points out the conditions under which a length of such a line can be considered as a transmission line with varying characteristic impedance. In the second section, the author deals with the synthesis of an incident reflection frequency function of a length of an inhomogeneous transmission line with unmatched terminals. The investigation of the designing problems of a transmission with prescribed parameters is based on the fact that the relation between the contour function

$$f(x) = - \frac{1}{2} \frac{Y'_0(x)}{Y_0(x)} = \frac{1}{2} \frac{d(\ln Z_0(x))}{dx} \quad (16)$$

where Y_0 is transmission line admittance and Z_0 , characteristic impedance

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Designing lengths of inhomogeneous transmission lines with prescribed reflection

of the transmission line, and the reflection frequency function of a length of an inhomogeneous transmission line, is analogous to the relation of irradiation function and directional characteristic of slot antennas, as described by Kovács and Solymár (Ref. 11: Acta Technica, 1956). The incident reflection frequency function is equal to one of the elements in the diagonal of the reflection matrix; the element, on the other hand, is equal to the Fourier transform of the contour function

$$f(x) = \mathcal{F}^{-1}\{h(\alpha)\} = \sum_{k=0}^n \frac{C_k}{2} \cos \frac{k\pi x}{L} : \quad (11)$$

Consequently, the synthesis is solved with the inversion of the Fourier transform. The inverse Fourier transform of the given reflection function is equal to the contour function from which the characteristic impedance

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Designing lengths of inhomogeneous transmission lines with prescribed reflection

function

$$Z_o(x) = \sqrt{Z_{o1} Z_{o2}} \exp \left\{ \frac{1}{2L} \left(\ln \frac{Z_{o2}}{Z_{o1}} \right) x + \sum_{k=1}^n \frac{C_k L}{k\pi} \sin \frac{k\pi x}{L} \right\} \quad (17)$$

is derived. The geometrical data of a length of an inhomogeneous transmission line can be computed from this equation where Z_{o1} and Z_{o2} are the characteristic impedances of lines to be connected. With the help of Eq. (17), the author and his associates designed and constructed a transition, with zero reflection at 3,600 and 3,800 Mc, for a microwave circuit of a broadband radio network. The standing wave ratio of the transition, as a function of frequency, was measured by conventional methods and the data are shown in Fig. 9. There are 9 figures and 17 references: 5 Soviet-bloc and 12 non-Soviet-bloc. The four most recent references to English-language publications read as follows: L. Solymar: On Higher Order

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Designing lengths of inhomogeneous transmission lines with prescribed reflection

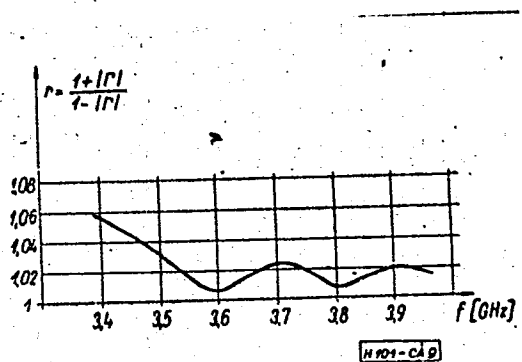
Approximations to the Solution on Nonuniform Transmission Lines, Proc. IRE, Nov., 1957; K. Matsumaru: Reflection Coefficient of E-Plane Tapered Rectangular Waveguide, IRE Transactions on MTT, April, 1958; K. Matsumaru: Reflection Coefficient of a Pyramidically Tapered Rectangular Waveguide, IRE Transactions on MTT, April, 1959; G. Reiter: Generalised Telegraphist's Equation for Waveguides of Varying Cross-section, Convention on Long-Distance Transmission by Waveguide, Jan., 1959, Proc. IEE, Part B. J

ASSOCIATION: Távközlési Kutató Intézet (Telecommunication Research Institute)

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Designing lengths of inhomogeneous transmission lines with prescribed reflection



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Fig. 9.

CSURGAY, A.

CSURGAI, A. [*Csurgay, A.*]; SENASHI, Sh. [*Szenasi, S.*]

Analysis of microwave noisy mixers. Acta techn Hung 42
no.1/3:171-179 '63.

1. Nauchno-issledovatel'skiy institut svyazi, Budapesht.

C.A. CSURI, I.

14

The graphic comparison of the chemical composition of medicinal waters of the Great Hungarian Plain. Hona Csuri, *Hidrol. Közlejt* 30, 62-5(1950).—The graphic method proposed by Maucha (C.A. 43, 1277c) was modified and a 24-angle diagram was introduced for illustrating the chem. compn. of waters. Besides the 9 groups of medicinal waters proposed by Than waters of alkali soils, characterized by Na_2CO_3 content, waters contg. 1 above 0.001 g./l. (iodic medicinal waters), and earthy-halide waters with a dominating Cl content and with about an equal content of Na and alk. earth metals. István Földi

CSUROS, Csaba, biologus

Single-cell parasites of our organism. Term tud kozl
8 no. 2: 77-80 F '64.

1. Baranya megyei Kozegeszsegugyi es Jarvanyugyi Allomas,
Pecs.



Therapy

HUNGARY

KAMARAS, Janos, Dr., and CSUROS, Eva, Dr., First Pediatric Clinic
at the University for Medical Sciences (Orvostudományi Egyetem, I.
Gyermecklinika) in Budapest (Director: GEGESI KISS, Pal, Dr.).

"Experiences with Steroid Therapy in Rheumatic Fever"

Budapest, Orvosi Hetilap, Vol 107, No 28, 10 Jul 1966, pp 1297-1300.

Abstract: The authors describe 237 cases of children suffering from rheumatic fever and carditis, respectively. All were afflicted for less than 30 days before treatment commenced, and all showed signs of the disease for the first time. The effects of therapy employing prednisolone + amidazophene and amidazophene alone were discussed with respect to total recovery and any cardiac changes that developed. The criteria for the classification of the cases were described and it was recommended that the treatment involving prednisolone be employed in all cases where there is a suspicion of carditis. 65 references, including 9 Hungarian, 1 Russian, 1 German, and 54 Western.

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CIA-RDP86-00513R00050942

On the problem of differential diagnosis of unilateral polycystic kidney. Magy. sebeszet 14 no.2:119-123 Ap '61.

1. A Budapesti Orvostudományi Egyetem Urológiai Klinika (Igazgató: dr. Babics Antal egyetemi tanár, akadémikus) és az I. sz. Gyermecklinika (Igazgató: dr. Gegesi Kiss Pal egyetemi tanár, akadémikus) közleménye.

(KIDNEY DISEASES diag)

LANCOS, Ferenc, dr.; CSUROS, Eva, dr.; KAMARAS, Janos, dr.; HALMAI, Zsuzsa, dr.

Ivemark syndrome associated with bilateral embolism of the common carotid artery. Gyermekgyógyászat 15 no.4:120-127 Ap'64.

1. A Budapesti Orvostudományi Egyetem I.sz. Gyermekklinikájának (igazgató: Gegesi Kiss, Pal, dr., akadémikus, egyetemi tanár) és a II.sz. Kórház Intézetének (igazgató: Haranghy, László, dr., akadémikus, egyetemi tanár) közleménye.

*

LANCOS, F.; CSUROS, Eva; KAMARAS, J.; HALMAI, Zs.

Ivemark's syndrome with bilateral embolism of the common carotid artery. Acta paediat. Acad. sci. Hung. 5 no.3:317-327 '64

1. I. Kinderklinik und II. Pathologisch-Anatomisches Institut
der Medizinischen Universität, Budapest.

KARVAS, J.; CSURCS, Eva

Steroid treatment in rheumatic fever. Acta paediat. acad. sci.
Hung. 6 no.3/4:359-365 1965.

1. First Department of Paediatrics, University Medical School,
Budapest. Submitted May 27, 1965.

CSUROS, Lajos

Gibraltar. Elet tud 18 no.40:1263-1267 6 0 '63.

1. "Elet es Tudomany" szerkeszto bizottsagi elnoke.

CSUROS, S.

Contributions to the study of macromycetes in the Harghita Mountains.
p. 12.

ANALELE ROMINA-SOVIETICE. SERIA AGRICULTURA
Vol. 10, No. 3, May/June 1956., Rumania

SOURCE: East European Accessions Lists, Vol. 5, No. 10, Library of Congress,
Oct. 1956.

CSUROs, ST.

COUNTRY : Rumania
CATEGORY : Meadow Cultivation. L
ABS. JOUR. : Ethiol., No. 1959, No. 15520
AUTHOR : Obrejaanu, Gr.; Mamas, M.; Velca, G.; Maxia, I.;
INST. : CSUROs, St.; Bannuila, I.; Axtor, Dora;
TITLE : Ciuj Arzibate, AS RPR
Increased Fertility in Natural Meadows of
Western Mountains (Rumania).
ORIG. PUB. : Studii si cercetari agron. Acad. RPR Fil. Cluj,
1957, 8, No.1-2, 99-116
ABSTRACT : No abstract.

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CARD: *Dresan, Viorica

CSUROS, S.

A review of Vegetation Studies in the Region of Orseg, Hungarian Part of the Eastern Slopes of the Alps by T. Pocs and others. p. 363.

Academia Republicii Populare Romine. Filiala Cluj. STUDII SI CERCETARI DE BIOLOGIE. Cluj, Rumania. Vol. 9, no. 2, 1958.

Monthly List of East European Accessions (EEAI) Vol. 8, no. 7, July 1959.

Uncl.

CSUROK, St.; RESMERITA, I.

The evolution process of prairies with *Festuca rubra* in Transylvania.
Studii cerc biol veget 13 no.2:223-232 '61.

(KEAI 10:11/12)

1. Catedra de botanica a Universitatii Babes-Bolyai si Sectia de
pasuni si finete a Sfatului Regional Cluj. Comunicare prezentata da
academician Emil Pop.

(*Festuca*)

CSUROS, Stefan

Botanical characteristics of the Transylvanian Plain. Analele
biol 17 no.5:117-127 Ag '63.

CSUROS, Stefan

Geobotanic studies on Pietrele Albe Mountain (Vladeasa Massif).
Studii cerc biol veget 15 no.1:71-90 '63.

RESMERITA, I.; NEMES, M.; CSUROS, St.

Stationary research on the grassy vegetation on the
Vladeasa-Micau Massif. Studii cerc biol veget 15
no.1:131-150 '63.

BECK, Mihaly; BITE, Pal; BRUCKNER, Gyozo; CSENTES, Jozsef; CSUROS, Zoltan;
DEAK, Gyula; ERDEY-GRUZ, Tibor; ERDEY, Laszlo; FABIAN, Pal;
FINALY, Istvan; FODOR, Gabor; FODORNE CSANYI, Pirooska;
GYORBIRO, Karoly; INZELT, Istvan; KUCSMAN, Arpad; NEUMANN, Erno;
PUNGOR, Erno; SCHNEER, Anna; SCHULEK, Elemer; SZABADVARY, Ferenc

Rules for the Hungarian chemical nomenclature and orthography.
Kem tud kozl MTA 17 no.1/4:1-292 '62.

1. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkeszto bizottsagi tagja (for Bruckner, Csuros, Laszlo Erdey, G.Fodor, and Schulek).
2. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" szerkesztoje (for Erdey-Gruz).
3. "A Magyar Tudomanyos Akademia Kemiai Tudomanyok Osztalyanak Kozlemenyei" technikai szerkesztoje (for Finaly).
4. Muvelodesugyi Miniszterium (for Csentes).
5. Magyar Tudomanyos Akademia Helyesitasi Bizottsage (for Fabian).
6. Nehezipari Miniszterium (for Neumann).

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCEDURES AND PROPERTIES INDEX																			
<div style="position: absolute; top: 10px; left: 10px; font-size: 24px; font-weight: bold;">15C</div> <div style="position: absolute; top: 10px; right: 10px; font-size: 24px; font-weight: bold;">a-3</div> <div style="position: absolute; top: 250px; left: 250px; border: 1px solid black; padding: 5px;"> <p>Action of nitroxy bromide on amino-acids. E. Carls (Magyar Chem. Vol. 1030, 36, 113--119; Chem. Zentr., 1930, I, 1443).—Substitution of the amino-group by bromine takes place only with α- and β-amino acids. The following substances are described: α-bromo-phenylalanine, m. p. 78–80°; α-bromo-β-benzyl acid, m. p. 60°; α-bromo-β-phenylalanine, m. p. 115°; α-bromo-β-phenylalanine, m. p. 100° (benzyl derivative, m. p. 61–62°).</p> </div>																			
ASS-SLA METALLURGICAL LITERATURE CLASSIFICATION																			
1900M CLASSIFICATION										1900M NOMENCLATURE									
1900M MET										1900M CHEM									
1900M MET										1900M CHEM									

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Synthesis of glucosides and compound sugars. ZOLTAN CACROS. *Magyar Chem. Folyoirat* 38, 44-55 (1932).—A short summary of theories of structure and description of the expts. of Haworth, Hudson and Zemplén, especially on the methods of investigation of structure by means of $Hg(OAc)_2$, $FeCl_3$ and HgI_2 . S. S. DE PINALY

Growth hormones. Zoltán Cacros. *Magyar Chemiai Folyoirat* 41, 150-63 (1935).—The characteristics of bays I, II and III and of auxin are discussed. S. S. de F.

ASB-33A METALLURGICAL LITERATURE CLASSIFICATION

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Condensation of urea with aldehydes. Zoltan Csizsar and Ferenc Vass. *Magyar Chem. Folyoirat* 43, 20-21 (1949). Urea 2 mols. and CH_3CHO 1 mol. in the presence of concd. AcOH on a steam bath gave a white substance in 79.7% yield, contg. 39.98-40.32% N. The fractions sol. and insol. in alc. have practically the same compn., the different solubilities are probably due to different degrees of polymerization. Two mols. urea with 3 mols. CH_3CHO both without any catalyst on heating and without heating in presence of concd. AcOH gave an 80.7% yield. Various other modifications of the method showed no practical differences. The condensation product was partially hydrolyzed by CH_3COOH or H_2SO_4 . Two mols. urea and 3 mols. AcH without catalysts or with 1% AcOH gave a white mass contg. 31.51-31.65% N. Treatment with 1% ammonia led to a formation of aldehyde resin instead of the wanted condensation product. The condensation product could not be decomposed. One mol. urea and 1 mol. vanillin in alc. soln. with 1% NaOH or 1 mol. urea and 2 mols. vanillin with 1% AcOH gave a yellow-brown mass contg. 5.02-6.67% N and 20.82-22.00% CH_3O . The product could partially be decomposed by dissolving it in 10% NaOH and adding HCl . One mol. urea and 1 mol. freshly distd. furfural gave on heating with 38% HCl an esbite-like black product contg. 23.48-23.87% N. One mol. urea in 2 mols. freshly distd. furfural gave after 1 day at room temp. a light-brown mass contg. 22.32-34% N. S. S. de Finlay

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CA

Chemistry of artificial materials (plastics). Z. Csurcs.
J. Hung. Chem. Soc. 1, No. 2, 1-12 (1940).—A
 review of such materials, viz., tech. org. macromol.
 plastics, their definition, classification, theoretical struc-
 ture, compn. and prepn.; the theory and examples of
 polycondensation and polymerization producing macro-
 mols. characteristic of such materials are included.
 B. C. P. A.

ASS-56A METALLURGICAL LITERATURE CLASSIFICATION

1ST AND 2ND INDEXES

3RD AND 4TH INDEXES

5TH AND 6TH INDEXES

7TH AND 8TH INDEXES

9TH AND 10TH INDEXES

11TH AND 12TH INDEXES

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99TH AND 100TH INDEXES

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1ST AND 2ND EDITIONS		PROCESSES AND PROPERTIES INDEX		3RD AND 4TH EDITIONS	
<p>ca</p> <p>Newer treatment methods in textile industries. Zoltan Csanics (Univ. Tech. Sci., Budapest, Hungary). <i>Kém. Tárja</i> 3, 108-12(1942).--Cellulose textiles should be treated with 4-12% of sizing material by immersion or brushing. This treatment increases somewhat the tearing strength of cotton and of cotton plus cottonized fiber; it does not affect the strength of rayon. The Tootal-Bronz does not affect the strength of rayon. The Tootal-Bronz were modified as to prevent deterioration of regenerated cellulose by acid-contg. catalysts at temps. above 100°, and to prevent loss of antiretard agent in ordinary washing. The antiretard resins are completely washed out of fabric in 5 min. at 70° in HCl soln. of pH 2 and in soda soln. of pH 11.</p>		<p>11</p>			
<p>ASS-563 RETALLURGICAL LITERATURE CLASSIFICATION</p>					
<p>1ST EDITION</p>		<p>2ND EDITION</p>		<p>3RD EDITION</p>	
<p>1ST EDITION</p>		<p>2ND EDITION</p>		<p>3RD EDITION</p>	

1ST AND 2ND CODES										3RD AND 4TH CODES									
PROCESSING AND PROPERTIES INDEX																			
<p>CA</p>										<p>Determination of N in nitro and nitroso compounds by the Kjeldahl method. Zoltan Czuros and Ernő Fodor-Kentzer. Magyar Chem. Folyóirat 48, 33-42 (1942); Chem. Zvest. 1943, 1, 845-6. The above procedure was found useful for the analysis of many nitro and nitroso compounds. W. T. H.</p>									
<p>ASB-514 METALLURGICAL LITERATURE CLASSIFICATION</p>										<p>FROM 00-100</p>									
<p>FROM 00-100</p>										<p>FROM 00-100</p>									

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CA

PROCESSES AND PROPERTIES INDEX

Viaceous rayon. Zoltán Csurgu and Sándor Grossmann.
 Hung. 132,165, Nov. 16, 1943. To prevent weakening of
 the fibers by H_2S evolved during the regeneration, oxidiz-
 ing agents (e.g., 1 g./l. $NaNO_3$) are added to the re-
 generating baths. István Pinyó

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

6-27-1943

13

Manufacture of phenoplastics from brown-coal tar-cresols. Zoltan Csuros and István Petri. *Magyar Chem. Folyóirat* 49, 11-26 (1943); *Chem. Zvesti.* 1943, II, 793—Rpts. showed that in general cresols from Hungarian brown-coal tar, purified according to Csuros and Zech (cf. C. A. 38, 6075), can be used satisfactorily in molding powders. The cresol, sold water-white and free of color, contains about 40-50% m-cresol. The chem. and mech. properties of the plastics made with this cresol meet the DIN-standards. In order to det. the min. amt. required to impart satisfactory mech., chem. and molding properties, the m-cresol content was varied from 20 to 75%. With 35% m-cresol, the chem. and mech. properties of the plastics were good but the molding properties were unsatisfactory. A satisfactory hardness could be obtained only by increasing the time of setting. Results were not judged by the abs. values obtained but, in order

to exclude expl. errors, by their relative values, i. e., values obtained by comparison with pure phenol plastics. The insufficient hardness of the cresol plastics can be improved by raising the content of m-cresol or by adding phenol, e. g., by using a cresol contg. 60% of the m-phenol, or by using 30% m-cresol with 20% phenol. The compd., or by using 30% m-cresol with 20% phenol. The molding powders of the type S meet the specifications excepting the hardness, like powders with cresols of other origins. This raises the manufg. costs somewhat. For origins. This raises the manufg. costs somewhat. For the manuf. of an excellent hardness the addn. of a trifunctional component is required. Satisfactory resols could be made with an NH₂ catalyst; these are a satisfactory substitute for cresols from bituminous coal in the manuf. of hardened paper, brake shoes, etc. A. K. Katerer

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<div style="display: flex; justify-content: space-between;"> ca 25 </div> <p>Alkali sensitivity of developed naphthol AS dyes containing nitro groups. <i>Zoltán Csizs</i> (Univ. Tech. Sci., Budapest, Hungary). <i>Magyar Chem. Folyóirat</i> 50, 25-33 (1944).—Most ice colors undergo color changes in the presence of dil. (even 5%) solns. of Na_2CO_3. The original color could be restored <i>in vitro</i> by heating the soln. with dil. acids, but this restoration was not successful with dyed textiles. The sensitivity of dyes to alkalis is connected with their structure, especially for compds. containing nitro groups located near azo groups. Compds. with nitro groups in the meta and para positions generally darken; <i>o</i>-nitro derivs. become lighter than the original color. The same observations were made on naphthol AS, naphthol AS-D, and naphthol AS-BO, even if bound to various diazotized bases. The results do not seem to affirm the quinoid structure. Probably the theory of Radulescu-Alexa is correct, the possible explanation being that nitro groups change the quinone hydrazones to a hydroxy azo compds., and simultaneously increase the alkali-binding capacity, as observed in the case of nitrated phenols. István Finály</p>																																																			
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<p><i>ca</i></p> <p>Corrosion of metal sleeves by a textile dye containing urea. Zoltán Csontos (Univ. Tech. Sci., Budapest, Hungary). <i>Mézőkém. Polytíral</i> 50, 33-7(1944).—Urea contained in a mixt. of dyes used for printing textiles was decompd. by urease to ammonia since the temp. of the factory was near to the optimal temp. of 30-38° of urease. The difficulty was overcome by adding 1-3 g. nipagin (Me p-hydroxybenzoate) to each kg. of dye mixt. The best method is to add the urea just before printing or to keep preps. in the refrigerator.</p> <p style="text-align: right;">István Finály</p>																										<i>25</i>																									
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C88

Derivatives of xanthates and aromatic nitro and amino compounds. Zoltán Csontos and István Ruszák. *Magyar Chem. Folyóirat* 50: 66-81 (1944).—Cellulose xanthate, dry or in soln., can be stabilized by salts of aromatic nitro sulfonic acids, e.g. $m\text{-O}_2\text{NC}_6\text{H}_4\text{SO}_3\text{H}$ (I). The structure of the products was proved by model expts. with simple xanthates which gave cryst. derivs. Alkali salts (except K) of I, $m\text{-O}_2\text{NC}_6\text{H}_4\text{NH}_2$, PhNO_2 , and $m\text{-O}_2\text{NC}_6\text{H}_4\text{NHAc}$, seem to be effective stabilizers. The products are esters of Ph- NHCSOR formed according to the equation $4\text{RNO}_2 + 6\text{R}'\text{OCS}_2\text{Na} + 3\text{H}_2\text{O} \rightarrow 4\text{R}'\text{NHCSOR}$ (II) + $3\text{Na}_2\text{S}_2\text{O}_3 + \text{CO}_2 + \text{CS}_2 + 2\text{H}'\text{OH}$. The following II were prepd. *Thiocarbonyl*, PhNHCSOR (R' given): *Me* (32.5% from MeOCS_2Na in 60% MeOH with PhNO_2), m. 97.5° (S 18.93, N 8.27, PhNH , 46.35-7.92%); *Et* (27.3% from EtOCS_2Na in 60% EtOH), m. 70° (S 18.1, N 7.62, PhNH , 45.55-7.00%); *iso-Bu* (35.7% from $\text{iso-BuOCS}_2\text{Na}$ in 90% EtOH), m. 78° (S 15.0, N 6.6%). *m-Sulfathio-carbonyl*: *Me* (67.6% from MeOCS_2Na and $m\text{-O}_2\text{NC}_6\text{H}_4\text{NH}_2$ in MeOH), contained S 24.0, N 5.18%; *Et* (83.1%), m. 270-8° (S 22.10, N 4.85, PhNH , 27.3-8.9%); *iso-Bu* (82.8%), m. 243-6° (S 19.9, N 4.39%); *glycerol* (34.5% from glycerol xanthate and III in aq. EtOH 4 hrs. at 35° and 20 mm.), amorphous, m. 153-5° (S 19.31, N 4.31%). Cellulose xanthate (7%) dild. with water and treated with 5-10% AcOH , the liberated H_2S removed, and the soln. treated with III 10 hrs. as above, gives a product (S 12.02, N 2.13, PhNH , 12.3-18.9%), identical with that obtained by Lillienfeld; the same compd. was obtained by treating 7% cellulose xanthate with NaOH , centrifuging, mixing with pptg. with satd. NaCl CS_2 , again dissolving in NaOH , and pptg. with satd. NaCl soln. *Me m-oxalamidethio-carbonyl*, m. 168° (N 12.39, S 14.37%), was obtained in 69.3% yield from MeOCS_2Na and $m\text{-O}_2\text{NC}_6\text{H}_4\text{NHAc}$ in 60% MeOH and AcOH heated 18 hrs. at 60°; *Et ester* (64.6%), m. 154° (N 11.61, S 13.64%). *Me m-nitrothiocarbonyl* (48% from MeOCS_2Na and $m\text{-O}_2\text{NC}_6\text{H}_4\text{NH}_2$ in 60% MeOH and a bit of

solid KOH heated 30 hrs. on a water bath), m. 119° (S 15.0, N 13.16%); *Et ester*, m. 115° (S 14.24, N 12.25%). The methods proposed by Orndorff and Richmond [*Am. Chem. J.* 22, 458 (1890)] and Hofmann [*Ber.* 3, 120 (1828)] were also tested. István Finály

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<p>Catalysis and catalysts. Zoltán Czura (Hung. Tech. Sci., Budapest, Hungary). Magyar Tsz. T. 41-5(1946). --No direct parallelism could be detected between the quantity of catalyst and the reaction velocity, but the existence of a max. and a min. could be proved. Variation of the amount of catalyst makes possible selective hydrogenation. István Finály</p>																			
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Selective catalytic hydrogenation. Z. Csikos (Tech. Inst., Budapest, Hungary). *Műgyárt. Közlönyek* 1947, 110-18; cf. C.A. 41, 1091. Selective or oriented catalysis may form geometrical isomers in a group of atoms in a mol. or in a component of a mixt. The selectivity of the catalyst is the result of a series of reactions or side reactions, since most chem. reactions proceed in several steps, and the catalyst accelerates only one of the intermediate reactions, or because the reaction will end in different stages depending upon the time of contact between the catalyst and the substrate. Cis and trans isomers were each hydrogenated at different rates by varying the amt. of Pd catalyst. The results do not agree with Paal's rule. The speed of hydrogenation is not a linear function of the amt. of catalyst, but reaches a max. and min. value with a definite amt. of catalyst. There is always an amt. of catalyst with which the cis modification is faster; but there is also an amt. with which the trans modification is faster. The results of the hydrogenation of the following acids were verified with fumaric and maleic acids; oleic and erucic acids; meconic and thalic acids; brassic and erucic acids; coumaric and cinnamic acids; cinnamic, coumaric, coumaric, and crotonic acids. The min. rate of hydrogenation of the trans modification always occurs with a smaller amt. of catalyst than that of the cis modification. Likewise, the same general observations regarding max. and min. amt. of catalyst and rates of hydrogenation were observed with the carbonyl bond in aldehydes and ketones. Also the optimum amt. of catalyst for the satn. of the carbonyl group is different than that required for the satn. of the ethylene group. PhCH:CHCHO was selectively hydrogenated at either point of unsatn. The cause of the selectivity of hydrogenation is the change in the course of

the max.-min. curve of the hydrogenation speed, which depends on the amt. of the catalyst. Further observations indicated that the carbonyl group of unsatd. aliphatic aldehydes and ketones cannot be hydrogenated but those of aromatic aldehydes and ketones are easily reduced. Crotonaldehyde absorbed 1 mole H_2 , PhCH:CHCHO and PhCH:CHCH_3 absorbed 2 moles. PhCH:CHCHO was reduced at atm. pressure with colloidal Pd or with Pd pptd. on animal charcoal. However, aliphatic unsatd. ketones and aldehydes (crotonaldehyde, tiglaldehyde, and methylheptenone) under the same conditions could not be reduced to the corresponding alcohols. Para substitution in aromatics accelerates reduction most and meta least. The selectivity with cis-trans isomers was observed with several catalysts, such as colloidal Pd, Pd on BaSO_4 , Pd and Pt on animal charcoal, and colloidal Pt. In various solvents, such as H_2O , AcOH , acetone, aniline, and toluene. H_2O was likewise selectively oxidized to either H_2O_2 or H_2O , using Pd on BaSO_4 as catalyst. Similar reduction selectivity was shown for the nitrile group. The catalytic hydrogenation of C_6H_5 and its derivs. likewise shift the hydrogenation in favor of the cis compd. PhC(COOH)_2 may either be reduced or reduced and decarboxylated. The duration of the dehydrogenation is a function of the speed const. and the order of reaction: $t = 1 - (1 - x)^{1/n}$. With various amts. of catalyst, not only the speed const. but also the order of reaction can assume a max.-min. curve. The order of reaction describes a value of 0 or a fraction. The speed shows a max. where the order most nearly approaches 0. The deviation from 0 of the order is caused by the hindering effect of the

reaction products, which restrains the further adsorption on the surface of the catalyst of the substance remaining unsatd., and thus is the more perceptible the smaller the difference between the adsorption of the compd. being satd. and the compd. already satd. Likewise, the amt. of adsorbed H_2 , referred to a unit of catalyst, also shows a min.-max. curve. The extreme values of the adsorption curve are the same as the extreme values of the hydrogenation-time curve with equal amt. of catalyst. While for the trans form the max. lies at the abscissa value of the adsorption max., for the cis form a max. is shown above the min. The position is reversed with Pt pptd. on animal charcoal. Thus the H_2 adsorption of the catalyst det. in advance the course of the speed curve. This involves a continuously changing 3-phase system where H_2 forms the gaseous phase, the solvent forms the liquid phase, and the catalyst forms the solid phase. A change in the amt. of catalyst changes the relationship between these 3 phases and between the substrates present in mol. state and the hydrogenation products.

R. R. Dunbar

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2

NEW PRINCIPLES IN CATALYSIS. Selectivity. Zolán
 Course. Magyar Kém. Lapja 3, 29-37(1948).—A sum-
 mary is given of literature data on selective catalysis.
 C's investigations consisted of selective hydrogenation of
 cinnamic aldehyde with colloidal Pd as catalyst and of
 Hall and a, p-, and m-hydroxybenzaldehyde in the pres-
 ence of Pd catalyst pptd. on animal charcoal. The
 Paal rule was found not to be valid. The amt. of cata-
 lyst can det. the compds. obtained in hydrogenation.
 (Lévy Pál)

ADD-31A METALLURGICAL LITERATURE CLASSIFICATION

CLASSIFICATION

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involvement on catalysts. VI. Mechanism and kinetics of hydrogenation of stereoisomeric ethylenecarboxylic acids. Zoltán Csúros, István Odázy, and Miklós Kepesti

(Univ. Tech. Sci., Budapest, Hung.). *Acta Chim. Hung.* 1, 22-43 (1951) (in German); cf. C.A. 44, 4706a. —Pl. animal charcoal prep'd. according to Gattermann-Wieland was used as the catalyst with maleic, fumaric, and cinnamic acids. The curves of hydrogenation velocity showed a max. and a min. with increased amounts of catalyst. A new method was developed for the detn. of the order of reaction, and the results proved that besides velocity of hydrogenation the order of reaction also changed, fluctuating between zero and a fractional one. This change is, however, not proportional to the amt. of catalyst but shows both a max. and a min. The velocity const. showed the highest values when the reaction order was approx. zero. With varying concns. the velocity const. increased with reduced concns. The product of the reaction has an inhibiting effect on the hydrogenation process, and its presence apparently changes the mechanism of the reaction. It was observed that the product of reaction retards both the absorption of the substance and the absorption of H₂. The differences observed in the hydrogenation velocities of maleic and fumaric acids are caused by differences in their absorptions.

VII. The role of the hydrogen pressure in the kinetics and mechanism of catalytic hydrogenation. Zoltán Csúros, István Odázy, and Tamás Négródi. *Ibid.* 108-81 (in English). —The role of H₂ was investigated in studies on the reaction kinetics and mechanism of the catalytic hydrogenation of ethylenic double bonds. When maleic and fumaric acids were hydrogenated in proportions of 1 mol. substrate to 1 mol. H₂, the reaction order and the rate const. as functions of the amt. of catalyst showed fluctuations following max.-min. curves. The overall order varied from 1.0 to 0 except where very small amts. of catalyst were used. The order with reference to H₂ was det'd. by hydrogenating 6.7 mol. substrate with 1 mol. H₂, in which case the distance between the extreme points on the curve became less. When the relative concns. of the reactants were reduced by diminishing the H₂ pressure to 0.5 of its original value resulting in a proportion of 1 mol. H₂ to 13.4 mol. maleic acid, the order of the reaction increased to 1.0. Succinic acid formed during the hydrogenation of maleic acid reduced the order of the reaction. The terminal value of order with reference to H₂ approached 2, and this proves that at least 1.5 and probably 2.0 mol. H₂ take part in this reaction. This justifies the Thomas theory and shows at the same time that the reaction has a free-radical mechanism. In the hydrogenation of 2-octene the max.-min. curve was replaced by a continuous function which agrees with results of preceding investigations and confirms the supposition that the max.-min. are caused by the adsorptive relations arising from the presence of a solvent. István Fényi

AcyI derivatives of cyclohexanone oxime, with special regard to the Beckmann rearrangement. Zoltán Csizsik, Kónstantin Zeeb, Gyula Dezy, and Etel Zoltányi, Univ. T-84 Budapest, Hungary. *Acta Chim. Hung.* 1, 66-83 (1951) (In German).—When cyclohexanone oxime (I) was treated with various acids of various concns. with no solvent or with CHCl_3 , $\text{CHCl}_2\text{-CHCl}_2$ or C_6H_6 , the mixt. cooled, or treated with 80% NaOH, filtered, the filtrate shaken out with CHCl_3 , the solvent evapor., and the residue dried, in some cases, the reaction products were either lactams, or ketones and lactams together, in 22-0-90-0% yields. By known methods the product of rearrangement can be obtained in high yields by the reaction of 0.5M mol. H_2SO_4 with 1.7mol. oxime at the lowest possible concn. of H_2SO_4 (65-3%), compared to 75% stated by other authors). P_2O_5 , PCl_5 , and HPO_3 are suitable for rearranging the oxime. SOCl_2 also effects the rearrangement; the rate of rearrangement depends, however, on the nature of the solvent used. New esters of I were prepd. by treating the *N*-deriv. (II) of I with various acid chlorides. They can be classified in 4 groups: (a) esters which on hydrolysis yield 1; (b) esters of bivalent acids behaving like the group (a), except for the diacid compound; (c) esters which give cyclohexanone and a peroxy acid or H_2O , when hydrolyzed; and (d) esters which give the Beckmann rearrangement. The following compounds were prepd.: 1) *acetic*, by 130°, and was obtained in 80.3% yield from I, $\text{C}_6\text{H}_{11}\text{N}$, CHCl_3 , and AcCl ; after shaking out with water and evapd. the CHCl_3 phase II (100%) was obtained by treating I with NaNH_2 in C_6H_6 soln.; *formic*, by 94.4%, with iso-BzCOCl, treating a suspension of II in C_6H_6 with iso-BzCOCl. Similar treatment of II with $(\text{COCl})_2$ gave 94.4% *I* oxalic, *I* succinic (68-80%) was obtained from II and SOCl_2 , maleic, II and SOCl_2 gave 85% *I* phthalic, m. 71°. Treatment of II with CS_2 and recovery of the product from MeCO gave the *dissulfide* compd. $(\text{C}_6\text{H}_{10}\text{N}_2\text{S}_2)$ (III), m. 65°. Hydrolysis of III gave a mixt. from which approx. equal amts. of I and cyclohexanone could be sep'd. as liquids (81.2%). BzCI, m. 58° was obtained from I in abs. $\text{C}_6\text{H}_5\text{N}$ with BzOI; BzCI , m. 58° was obtained from I in abs. $\text{C}_6\text{H}_5\text{N}$ with HOOC ; BzCI , m. 51° was obtained from II with PhSOCl . *I* 3-oxobutanoic-sulfonic (55.7%), m. 51°, was obtained from II with 2-CaHSO $_4$. *I* p-toluenesulfonic (97.5%), m. 60°, was obtained from II with p-MeC $_6$ H $_4$ SO $_3$ Cl. *I* o-methanesulfonic was obtained in liquid form from II with o-MeC $_6$ H $_4$ SO $_3$ Cl. II treated with phosy chloride, gave 65.6% *I* phosoric (IV), m. 83°. IV boiled 5 hrs. in (CH_2Cl_2) rearranged to the *N*-phosyl lactam, m. 175°.

Issván Fényi

CA

10

The sulfuric acid ester salt of cyclohexanone oxime. Zoltán Csúros, Konstantin Zech, and Sára Zech (Univ. Tech. Sci., Budapest, Hung.). *Acta Chim. Hung.* 1, 83-93 (1951) (In German); cf. preceding abstr. — The structure of the H_2SO_4 ester salt of cyclohexanone oxime (I) prepd. from I and N_2SO_4H

(IA) or from cyclohexanone and H_2NOSO_2H was examd. Cyclohexanone with sulfoperamidic acid in the presence of an equiv. amt. of alkali as described by Knoll (cf. Ger. patent 540,409, C.A. 26, 3263), gave a product identical to that obtained by treating I with IA. Catalytic hydrogenation of this product proved that it is not a deriv. of ϵ -aminocaproic

acid lactam but a H_2SO_4 ester salt (II) of I, $(CH_2)_5C:NOSO_2K$. Catalytic hydrogenation of II gave cyclohexylamine and dicyclohexylamine. II was surprisingly sensitive to alkali, which is unusual for a ketoxime H_2SO_4 ester salt. This is caused by the cleavage which is effected not only by mineral acids but also by alkalis with a simultaneous rearrangement. The cleavage products were ϵ -leucine lactam and $KHSO_4$. The H_2SO_4 group is cleaved in this case, differing from other ketoxime H_2SO_4 ester salts, not as a salt of sulfoperamidic acid but as a H_2SO_4 salt, leaving N on the residue. István Finály

CSUROS Z.

181725

HUNGARY/Chemistry - Catalysts

1951

"Role of the Hydrogen Pressure in the Kinetics and Mechanism of Catalytic Hydrogenations," in English, Z. Csuros, I. Geczy, T. Nogrady, ORG Chem Tech Inst, U Tech Sci, Budapest

"Acta Chimica Acad Sci Hungaricae" Vol I, No 1, pp 168-181

Examd kinetics and mech of catalytic hydrogenation reaction of double bonds with respect to hydrogen pressure, and retarding effect of reaction product under these conditions. Modified course of max-min curve and studied by systematic variation

181725

HUNGARY/Chemistry - Catalysts (Contd)

1951

of substrate concn and by hydrogenation without solvents. In the case of octene-2 without solvent, max-min curve disappeared and gave place to monotonous function.

181725

CSUROS, Z.

HUNG.

1. Catalysts. X. Rate and selectivity of hydrogenation of sunflower oil in the presence of nickel formate catalyst. Zoltan Csuros, István Gábor, and Gyula Szabó (Tech. Univ. Budapest), *Acta Chim. Acad. Sci. Hung.* 2, 53-54 (1952) (in English); cf. C.A. 48, 2005f. — A soln. of 110 g. Na_2CO_3 in a little H_2O was added slowly to a boiling soln. of 260 g. $\text{NiSO}_4 \cdot 7\text{H}_2\text{O}$ in 500 g. H_2O ; the resulting ppt., after washing with H_2O until neutral, was dissolved in 90-100 g. HCO_2H , boiled, and filtered, giving 181 g. (85%) green cryst. $\text{Ni}(\text{HCO}_2)_2$ (I) after drying at 100° . Treatment of a Ni salt with HCO_2Na gave I in 84% yield. The catalytic activity of I was not reduced by the presence of an Fe impurity but was increased by reduction of the particle size to 10,000 mesh/sq. in. Heating of a 1:1 suspension of I and refined sunflower oil (II) at $245-50^\circ$ for 15-20 min. in a vacuum gave a black oily suspension of Ni which was washed 3 times by decantation with five-fold its vol. of petr. ether; removal of all traces of solvent in a vacuum desiccator (to which air is then admitted slowly over 0.5 day to prevent destruction of the Ni during glowing) and sifting of the ppt. through a 10,000 mesh/sq. in. sieve, gave a catalyst (III) consisting of 60% Ni in adsorbed oil. III is stable when stored in a glass-stoppered bottle. The rate of hydrogenation of II with 0.4% (optimum amt.) III increased by raising the temp. from 160° to 200° and declined above 200° . An induction period was present at $120-40^\circ$. By controlling the rate of hydrogenation, II could be selectively reduced so that only 1 of the double bonds in the linoleic acid became satd. Selectivity increased with larger quantities of III up to 0.8% and with a rise in temp. up to 220° ; above this temp. and concn., selectivity declined. M. Jacobson

CSÜROS, ZOLTAN

Catalysts. 31. The role of autooxidation processes in the formation of films of drum oils. Zoltan Cseres, Zoltan Hayos, and Gyula Deak. Chim. Ind. (Milan), 1963, 45, 1130-1131; Chim. Acta Slov. Hung., 2, 214-20 (1959); Environ. Sci. A, 49, 1130-31. -- Absorption of O by H_2R in C_6H_6 in the presence of Co and Mn dries increased with increased amounts of drier. In experiments with Hungarian based oil technical grade, on strips of Japanese paper with 0.02 and 0.04 mm thickness driers were used in amounts of 0.02, 0.04, and 0.08 g. Co^{2+} and 0.02, 0.04, and 0.08 g. Mn^{2+} per g. unoxidized oil; the rate of autooxidation was measured and decreased. The peak was attained at 10 min. after the start of the presence of Co^{2+} in the S_0 and was 1.6 times higher than with Mn^{2+} in the S_0 up to 10 min. After 10 min. the maximum percentage no longer appeared to be associated with the drier amount. When heated at 160°C. for 10 min. the S_0 in air for periods up to 6 hrs. the changes of peroxide no. were insignificant. Under ultraviolet irradiation the no. increased sharply to a max. (650 ml. 0.02 N $\text{Na}_2\text{S}_2\text{O}_3$ for 1 g. oil); a simple heating to 140°C. led to max. lower than

[illegible]

Z. CSUROS, J. FODOR, Z. HAJOS.

"Investigations on catalysts. XII. Effect of ion exchangers in esterification" p. 459
(ACTA CHIMICA ACADEMIAE SCIENTIARUM HUNGARICAE, Vol. 2, No. 4, 1952, Budapest, Hungary)

SO: Monthly List of East European Accessions, L.C., Vol. 2 No. 7, July 1953, Uncl.

ZOLTAN CSUROS

Hungary/ Organic Chemistry - Theoretical and general questions
on organic chemistry

E-1

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11602

Author : XI. Csuros Zoltan, Hajos Zoltan, Deak Gyula.
XII. Csuros Zoltan, Fodor Jozsef, Hajos Zoltan.

Title : Investigation of Catalytic Reactions. XI. Role of Autoxidation Processes in Formation of Drying Oil Films. XII. Effect of Ion-Exchange Resins on Esterification Reaction.

Orig Pub : Katalizatoros vizsgalatok. XI. Autoxidacios folyamatok szerepe szarado olaj filmek kialakulasaban. XII. Ioncserelek hatasa eszterezesre. Magyar tudoman. akad. kem. tudoman, osztalyanak kozlemenyei 1953, 3, No 4, 469-485; 501-513 (Hungarian)

Abstract : Communication XI. It was ascertained that on autoxidation of C_6H_5CHO in $CHCl_3$ rate of O_2 absorption increases in the case of addition of the siccatives, naphthenates of Co (I) and Mn (II). In the case of large amounts of I or II amount of absorbed O_2 is greater. After absorption of a certain amount of O_2 the system reaches an equilibrium and further absorption of O_2 is not observed. On autoxidation of linseed

Card 1/5

Hungary/ Organic Chemistry - Theoretical and general questions
on organic chemistry

E-1

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11602

oil (LO) in the presence of I or II peroxide value (PV) rises at first and then decreases. In the presence of I maximum PV is reached sooner than in presence of II. Highest value of maximum is associated with the use of definite, optimal, amount of siccative; in the presence of lesser or greater amount of siccative value of PV maximum decreases. Investigated were the variations of PV of LO on irradiation with diffused light, ultraviolet radiations and on heating at 140°. In the first mentioned case a slow linear increase of PV is observed. Under the action of ultraviolet radiations PV increases faster and reaches a definite maximum level. On heating an analogous maximum is reached still faster after which PV drops rapidly; in this case maximum value is lower than on irradiation with ultraviolet. Boiled oil prepared with the use of II, shows on irradiation with diffused light an increase in PV only after a prolonged induction period, after which PV rises sharply to maximum level and then decreases. On exposure to ultraviolet radiations PV of oil increases sharply to a maximum the value of which is higher than in the preceding instance; following the maximum PV drops

Card 2/5

Hungary/ Organic Chemistry - Theoretical and general questions
on organic chemistry

E-1

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11602

very rapidly. In this case the maximum is reached more rapidly than in the case of unboiled LO, and the process of decomposition of peroxides also occurs more rapidly than with LO. On heating at 140° very rapid decomposition of peroxides is observed in the boiled oil under study. In boiled oil PV maximum is attained considerably sooner than in the case of drying oil prepared in the cold with the same siccative as is used in the boiled oil, but in the latter instance the maximum PV level is much lower. During the initial period of drying of LO the chain process of autoxidation is playing a substantial part. As the drying progresses autoxidation becomes of subordinate importance and decomposition of peroxides sets in, which is probably associated with occurrence of bond formation between activated molecules of the oil. This confirms the previously proposed mechanism of drying (Powers P.O. et al., Ind. Eng. Chem., 1951, 33, 1257). As concerns the mechanism of action of the siccatives, it was ascertained that they are capable of catalyzing the formation and the decomposition of peroxides.

Card 3/5

Hungary/ Organic Chemistry - Theoretical and general questions
on organic chemistry

E-1

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11602

Communication XII. The reaction of phthalic anhydride with butanol was utilized to investigate the effect upon the reaction of esterification, of fractions of different degree of dispersion, of a cation exchange resin of phenol sulfonic acid type, Wofatit KS. Activity of catalyst (amount of acid group titrated with KOH) increases with increasing degree of dispersion up to a certain limit after which increased degree of dispersion produces no appreciable effect on the activity. Velocity of reaction increases with increasing amount of catalyst up to a certain optimal amount; on use of greater amounts of catalyst velocity of the reaction decreases. On increase of the degree of dispersion of the catalyst its catalytic action (terminal degree of conversion) decreases except for those instances when a highly dispersed catalyst is subjected to activation after comminution. By means of ion-exchange resins it is possible to attain the same extent of conversion as with an equimolecular amount of H_2SO_4 , but more slowly than with H_2SO_4 .

Card 4/5

Hungary/ Organic Chemistry - Theoretical and general questions
on organic chemistry

E-1

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 11602

To activate the catalyst it is treated with dilute H_2SO_4 while stirring ($\sim 20^\circ$). Drying of the catalyst (60°) to constant weight has a detrimental effect on its catalytic properties, therefore it is used in air-dry condition.
Communication X see RZhKhim, 1955, 28809.

Card 5/5

CSUROS, Z.

Hungarian Technical Abst.
Vol. 6 No. 1
1954

8-31-54
[Signature]

541.126.04
3. Autoxidation processes - Antioxidants polymers -
L. Z. Csizs, (Hungarian Journal of Chemistry - Magyar
Kémiai Folyóirat - Vol. 59, 1953, No. 3, pp. 83-91, 10
figs., 4 tabs.)

It was found (contrary to former data in literature) that the autoxidation rate of ascorbic acid in alkaline media is higher and proceeds further than in acid media, i. e. oxidation is promoted by a higher pH. The effect depends on the bases used, i. e. if the same pH is adjusted by potassium hydroxide, sodium hydroxide or ammonium hydroxide the effect is different. The pH decreases during the reaction, except between pH 2.0 and 5.5, and the higher the starting pH the more it decreases. By using buffer systems which counterbalance the acidification process, the rate of oxidation increases. Certain compounds, themselves autoxidizable, e. g. hydroquinone, increase the oxidation rate by oxygen transport. Materials with a large specific surface (e. g. charcoal, carbon black) exhibit the same properties probably due to their ability to absorb oxygen. In former experiments on the oxidation of benzaldehyde in a 10% carbon tetrachloride solution with palladium as catalyst on barium sulfate, it was observed that a certain amount of catalyst could be replaced by an inactive carrier without a change in the reaction rate. This "carrier effect" was investigated in different solvents and on benzaldehyde using different catalysts (e. g. palladium on barium sulfate, cupric oxide on kaolin, etc.). It was concluded that the "carrier effect" is observable not only in connection with precious metal catalysts but in the case of metal oxides (e. g. cupric oxide) as well.

Csürös, Z.

Investigation of catalysis. XIII. Autoxidation of ascorbic acid as a function of pH values. Z. Csürös and L. Petró (*Acta chim. hung.*, 1955, 7, 199—222).—The oxidation of ascorbic acid (I) is first-order: the curve of its rate constant as a function of initial pH values shows two sharp max. at pH 5 (1 mol. of alkali) and 11.5 (2 mol.), but the use of NaOH, KOH, aq. NH₃ and other bases (e.g., tartrate, phosphate and acetate) leads to different results with each base. In general, the O uptake increases with initial pH values up to 10 mol. of alkali, and proceeds to 2 atoms of O per mol. of I. At the pH 5 max., dehydroascorbic acid is the only product. Apart from this narrow region the oxidation is irreversible and proceeds by several simultaneous routes. The O uptake of dehydroascorbic acid is low in acid and alkaline solutions, and changes produced are mainly degradations. (28 references.) T. P. McLAUGHLIN.

Mc
CH

①

So: Monthly List of EEAR, LC, Vol. 5,
No. 3 March '56

Csüros, Z.

Investigations on catalysts. XV. Polymerization and catalytic hydrogenation as concurrent reactions. 1. Gérvy and Szilvák (Acta chim. hung., 1955, 8, 283-294). Both the methyl acrylate and the acrylate can be polymerized with catalytically activated H at 18° in absence of per-compounds, but acrylonitrile, styrene and vinyl acetate can be polymerized similarly only in presence of a per-compound (e.g., 0.05 g. of K₂S₂O₈ per g. of ester). Hydrogenation and polymerization are probably concurrent reactions in these redox systems, the concn. of H⁺ at the catalyst surface and the relative rates of hydrogenation and polymerization determining which reaction is dominant. In presence of very high concn. of H⁺, hydrogenation occurs, but polymerization takes place at low concn. of H⁺. The formation and influence of free radicals of H in these reactions is discussed. W. J. BAKER.

3
M.A. YOUTZ
2 copies

PM

CSUROS, Z.: GECZY, I.

Magyar Kemikusok Lapja - Vol. 10, no. 5, May 1955

O-cresol as the new raw material of our chemical industry. p. 153.

SO: Monthly list of East European Accessions, (EEAL), LC, Vol. 4, No. 9, Sept. 1955
Uncl.

CSUROS, Z. (Prof., Dr.)

Hungary

Neue Entwicklung auf dem Gebiet der Theorie und Praxis der Hochpolymeren

(Hauptjahrestagung 1956 der Chemischen Gesellschaft in der Deutschen Demokratischen Republik).

Aus dem Tagungsprogramm - Nachmittags: Gruppe C:

Prof., Dr. Z. CSUROS, Budapest, "Polymerization und katalytische Hydrierung als konkurrente Reaktionen."

SOURCE: Plaste und Kautschuk, October 1956, Unclassified.

CSUROS, Z.; PETRO, J.

CSUROS, Z.; PETRO, J. Examinations by catalysts. XVI. Autoxidation of ascorbic acid in the starting pH function by various alkalies. P. 137.

Vol. 8, No. 1, 1956

KOZLEMENYEI

SCIENCE

Budapest, Hungary

So: East European Accession, Vol. 6, No. 2, Feb. 1957

Source 2

CSUROS, Z.

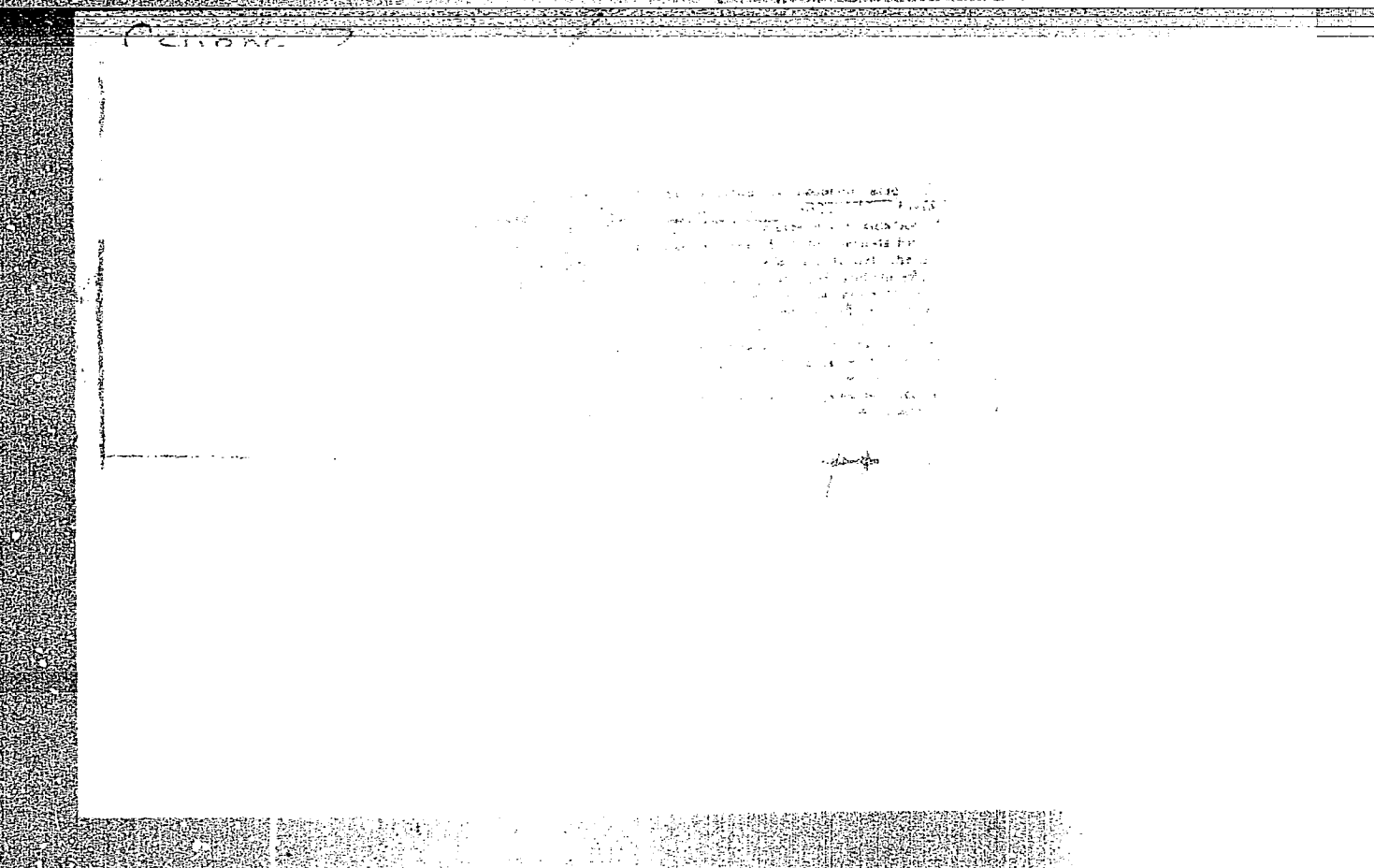
CSUROS, Z. - Geza Zemplen (1883-1956). p. 225.
Vol. 11, no. 8, Aug. 1956
Magyar Kemikusok Lapja, Budapest, Hungary

SOURCE: East European Accessions List (EEAL) Vol. 6, No. 4--April 1957

CSUROS, Z.

CSUROS, Z. - Lajos Kovacs Manyag zsebkönyv (Handbook on Plastic Materials);
a book review, p. 259
Vol. 11, no. 8, Aug. 1956
MAGYAR KEMIKUSOK LAPJA - Budapest, Hungary

SOURCE: East European Accessions List (EEAL) Vol. 6, No. 4--- April 1957



HUNGARY / Organic Chemistry. Natural Substances and
Their Synthetic Analogues.

G-3

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57600.

Author : Csuros Z., Petro J.

Inst : Not given.

Title : Investigations of Catalysis. XVII. Autooxidation
of Ascorbic Acid as a Function of Temperature and
Initial pH.

Orig Pub: Magyar tud. akad. kem. tud. oszt. kosl., 1957,
No 1, 43-60.

Abstract: Effect of temperature on the addition of oxygen
(AO) to ascorbic acid (I) was investigated. At a
pH < 3, AO increases to a greater degree with in-
creasing temperature than in neutral media (at pH
of 4-8) or in alkaline media (pH > 8). At a 0.4
pH (acidified with HCl) and at 60°, 5 times more

Card 1/5

72

HUNGARY / Organic Chemistry. Natural Substances and
Their Synthetic Analogues.

G-3

Abs Jour: Ref Zhur-Khimiya, 1958, No 17, 57600.

Abstract: O₂ goes into the combination than it does at 30°
and with pH varying from 4-8 and > 8. With the
increase in temperature from 30° to 60° the AO
increases only by 20-30%. In alkaline media the
AO proceeds faster, with the equilibrium being es-
tablished in 1-2 hours. In an acid medium the
equilibrium is not reached in 3 hours. At a pH of
4-6 and at a temperature of 40° and higher, the
increase in pH, as the result of reaction, is not
noticed. At 50° and higher, in all the cases, pH
tends to decrease toward the end. The highest AO
was noticed at 90°. Three AO maxima occur at 0.5,

Card 2/5

Distr: hE2c(j)

4. A rheological study on high-molecular materials. 1.
(In German) Z. Csüros, T. Geczy, M. Grosz-
mann, K. Várkonyi. *Periodica Polytechnica, Chemical
Engineering*, Vol. 1, 1957, No. 2, pp. 105-129, 13 figs.,
13 tabs.

The flow curves, pour points, elasticity and hardness
of typographical ductor rolls made of gelatine and starch
were determined by means of a Hoeppler consistometer.
It was found that the Herschel-Bulkley equation

$$\left(\frac{\partial \gamma}{\partial t} = A(\tau - \tau_0)^n\right)$$

is valid for the quasi-viscous curves
and flow curves of both materials where $n < 1$ for gelatine-
based roll materials and $n > 1$ for starch-based substances.
The starch material stands up to compressive stresses
in practice better than the gelatinous substance and
the consistency of the former changes considerably less
with the increase of temperature. The elastic properties
of the starch-containing material are more favourable
than those of the gelatinous substance since the former
undergoes greater immediate and lesser delayed and per-
manent deformation. The starch-containing roll material
is plasto-elastic below 40°C and visco-elastic above this
temperature. The corresponding characteristic tempera-
ture for the gelatinous material is 30°C.

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Distr: 4E2c/4E3d

59. Investigation on the use of Raney nickel catalysts.
(In English) Z. Cašarsa, J. Petro, J. Vörös
Periodica Polytechnica, Chemical Engineering, Vol. 1,
1957, No. 3, pp. 153—185, 29 figs., 2 tabs.

The change of the hydrogenating activity of Raney nickel was studied by varying the conditions of the preparation (dissolution, time and temperature of after-treatment) using acetone, acetophenone, benzophenone, eugenol and veratrol as model compounds. The aromatic ring of veratrol was hydrogenated at 30 atm and 160°C, the other compounds being treated at atmospheric pressure and room temperature. It was found that the activity of the catalyst and the extent of hydrogenation depends greatly on the conditions of preparation, the degree of the effect differing for each model. The effect of various added materials, such as organic and inorganic bases and high-molecular N-containing organic compounds, was investigated as well. The greatest increase of activity was produced with dimethyl aniline used in amounts of 0.1 mol referred to the substrate. Triethylamine was also found effective. Inorganic bases and high-molecular organic materials had either no effect or an unfavourable effect.

HUNGARY / Physical Chemistry. Kinetics. Combustion. B-9
Explosives. Topochemistry. Catalysis.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 76756.

Author : ~~Csueross, Z.~~ Geczy, I., and Keresztely, T.
Inst : Not given.
Title : Catalytic Studies. VIII. Change in the Reaction
Order in the Hydrogenated Compound During Cata-
lytic Hydrogenation.

Orig Pub: Magyar Tud Akad Kem Tud Oszt Koezl, 9, No 2,
195-205 (1957) (in Hungarian).

Abstract: On the basis of previously published data on the
catalytic hydrogenation of cinnamic, maleic, and
fumaric acid, the authors arrived at the conclu-
sion that in the course of the hydrogenation the
order of the reaction in the hydrogenated com-

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HUNGARY / Physical Chemistry. Kinetics. Combustion. B-9
Explosives. Topochemistry. Catalysis.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 76756.

Abstract: pound increases in the majority of cases from 0
to 0.6. In isolated cases when very small amounts
of catalyst are used the order of the reaction
does not change. For communication XVII see
RZhKhim 1958, 57600.

Card 2/2

HUNGARY / High Molecular Chemistry.

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Ibs Jour : Ref Zhur - Khim., No 3, 1959, 10562

duration of stirring, and time of storage, on properties of pastes prepared from poly(vinyl chloride) powder and different plasticizers of the dialkyl phthalate series. For this purpose a determination was made of the viscosity curves of the preparations under study. Shear stress was measured in the interval 10-300 g/cm², displacement rate varied between 0.1-5 cm/sec. Flow velocity of the pastes under study is not proportional to shear stress. Viscosity of pastes obtained on vigorous stirring is higher than that of corresponding pastes prepared with gentle stirring. This difference is enhanced with increasing content of poly(vinyl chloride) in the paste. Viscosity is also increased with increased duration of stirring. After stirring for 50-60 minutes, the increase in viscosity reaches maximum value. Change in

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HUNGARY / High Molecular Chemistry.

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Ibs Jour : Ref Zhur - Khim., No 3, 1959, No 10562

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viscosity on standing depends on composition of paste and the duration of stirring at the time of preparation. Duration of stirring at the time of preparation. Duration of stirring has a considerably greater effect than length of storage of the paste.

III. An Ostwald viscosimeter was used to determine characteristic viscosity $[\eta]$ of some samples of polyethylene of known molecular weight M , in paraffin oil (paraffin hydrocarbons containing 20-21 C-atoms) at 85° and 140°. The following correlations were found: $[\eta] = 2.898 \cdot 10^{-4} M^{0.7762}$ at 85°; $[\eta] = 2.766 \cdot 10^{-3} M^{0.7762}$ at 140°. Part I, see RZhKhim, 1958, 72806. -- S. Zelilman.

Card 3/3

251

Distr: 4E2c(j)

147. Experiments for the production of after-chlorinated polyvinyl chloride in Hungary. (In German) Z. Csürös, M. Groszmann, H. Zaffa. Periodica Polytechnica, Chemical Engineering, Vol. 2, 1958, No. 1, pp. 59-63, 3 tabs.

The conditions of chlorination of polyvinyl chloride were studied in different solvents and with different catalysts as a function of time. The reaction was found to take place quicker in higher-boiling solvents, however a temperature limit is set by the danger of the decomposition of the product. Several catalysts and catalyst mixtures were tried besides those known from literature with which the rate of the chlorination reaction could be favourably influenced. Special care had to be taken during the experiments to avoid excessive molecular degradation. In order to obtain products of uniform properties the temperature, concentration and time of chlorination must be closely controlled. The primary purpose of the experiments was to produce a technically useful adhesive material, at the same time interesting observations were made in connection with the catalysis.

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✓ Polymerization through catalytically activated hydrogen.
Z. Tsauros and T. Géczy (Tech. Univ., Budapest, Hung.).
Periodica Polytech. 2, 65-88(1958).—Some unsatd. compds.
were polymerized in the presence of H and a hydrogenation
catalyst. Thus, Me methacrylate, Me acrylate, and Et
acrylate were polymerized in MeOH in the presence of H
and Pd on BaSO₄. The H· radicals that were formed
initiated the polymerization. Acrylonitrile, styrene, and
vinyl acetate required a peroxide in addn. to the activated
H. The rates of these reactions were measured.

Millard Maienthal

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Distr: 4E3d/4E2c(j)

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Distr: 4E20(j)/4E3d

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Investigations by use of boron trifluoride and its complexes. Csuros and Gy. Deák (Polytech. Univ., Budapest, Hung.). *Periodica Polytech.* 2, 135-43 (1958).—The speed of anomerization of sugar esters, catalyzed by BF_3 (gas), $\text{BF}_3(\text{AcOH})_2$ (I), or $\text{BF}_3\text{Et}_2\text{O}$ (II) in the solvents Ac_2O , AcOH , CHCl_3 , and mixts. of Ac_2O - AcOH depended on the catalyst concn. and the types of solvent. The reaction const. $(k_1 + k_2) \times 10^4/\text{min.}$ for β -pentaacetylglucose (0.1 mol./l.) in Ac_2O soln. are for catalyst I = 206.0; II = 70.5; in AcOH for I = 8.70; II = 5.20. The high acidity of the Ac_2O solvent is responsible for the high reaction const. The relative basicities of AcOH and Et_2O (introduced with catalyst complex) as compared with BF_3 are believed to hold a portion of the BF_3 bound and thus decrease the effective catalyst concn. Reaction const. for anomerization of pentaacetylglucose at 25.3° with 0.125 mol./l. of I for various molar concns. of $\text{Ac}_2\text{O}/(\text{AcOH})_2$ are: 0.0000/0.2020 = 3.20; 0.0031/0.1142 = 3.52; 0.0002/0.0090 = 7.85; 0.1030/0.0475 = 17.40; 0.1160/0.0260 = 38.70; 0.1290/0.0000 = 206.00. Roland E. Kreibitz

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Z. CSUROS

Distr: 43-c(j)

Chlorination of poly(vinyl chloride). II. Z. Csuros, M. Groszmann, and B. Zsuffa (Polytech. Univ., Budapest, Hung.). *Periodica Polytech.* 2, 183-7(1958). Chlorination of poly(vinyl chloride) (I) (55% Cl) in CCl_4 , $\text{C}_2\text{H}_5\text{Cl}$, $\text{C}_6\text{H}_5\text{Cl}$, and PhCl in the presence of TiCl_4 , PCl_5 , SbCl_5 , and BF_3 gave the highest chlorination with SbCl_5 in $\text{C}_2\text{H}_5\text{Cl}$ at $130-6^\circ$. In CCl_4 at 78° the end product had % Cl with PCl_5 = 57.0 (12 hrs.); SbCl_5 = 58.2 (8 hrs.); TiCl_4 = 56.7 (8 hrs.). In $\text{C}_2\text{H}_5\text{Cl}$ at $82-4^\circ$ with 1% TiCl_4 = 58.9, 3% = 62.2, 5% = 60.3% Cl; with 3% PCl_5 = 59.8, with 5% PCl_5 = 61; with 5% SbCl_5 = 60.0, 3% BF_3 = 60.0% Cl in end product. In $\text{C}_2\text{H}_5\text{Cl}$ at $130-6^\circ$ with 1% TiCl_4 = 60.9, 3% TiCl_4 = 60.7-2.5, 5% TiCl_4 = 61.7% Cl; with 3% SbCl_5 = 61.3, 5% SbCl_5 = 64.7; with 5% PCl_5 = 61.0% Cl. Chlorination in PhCl at $130-2^\circ$ for 4.5 hrs. gave with 1% TiCl_4 = 58.2, 3% TiCl_4 = 59.3, 5% TiCl_4 = 59.4% Cl; with 5% SbCl_5 = 58.3, 5% BF_3 = 59.3% Cl in final I. A catalyst mixt. of TiCl_4 and BF_3 (anisole complex) in PhCl ($130-2^\circ$, 4.5 hrs.) gave I with 60.2% Cl, in $\text{C}_2\text{H}_5\text{Cl}$ ($82-4^\circ$, 5 hrs.) 58.6%, in $\text{C}_2\text{H}_5\text{Cl}$ ($130-3^\circ$, 3.5 hrs.) 60.7%. Mol. wts. of the I obtained were 29,000-53,000. No correlation was found between Cl content and mol. wt. Catalyst concn. in excess of 3% did not improve the rate of chlorination.

Roland E. Kriebitz

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HUNGARY / Physical Chemistry--Kinetics. Combustion. B-9
Explosions. Topochemistry. Catalysis.

Abs Jour : Referat Zhur--Khimiya, No. 11, 1959, 37927

Author : Csueres, Z.; Geczy, I.; and Czuffa, B.
Inst : Hungarian Academy of Sciences
Title : Investigation of Catalysts. XXI. Catalytic
Hydrogenation and Polymerization Processes as
Competing Reactions. III. Kinetics and Mechanism of the Catalytic Redox Polymerization of
Acrylonitrile.

Orig Pub : Magyar Tud Akad Kem Tud Oszt Koezl, 2, No. 4,
423-432 (1958) (in Hungarian); Makromolek Chem,
27, No. 3, 180-191 (1958) (in German)

Abstract : The authors have studied the polymerization of
aqueous acrylonitrile solutions under an atmo-

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HUNGARY / Physical Chemistry--Kinetics. Combustion. B-9
Explosions. Topochemistry. Catalysis.

Abs Jour : Referat Zhur--Khimiya, No. 11, 1959, 37927

sphere of H_2 in the presence of $K_2S_2O_8$ (I), using Pd catalysts precipitated on $BaSO_4$. It is shown that the rate of polymerization is proportional to the square root of the concentration of I and that with increasing monomer concentration, the rate of polymerization passes through a maximum. The degree of polymerization depends linearly on the time. The presence of an induction period has been established; the length of that period depends on the concentration of I. The kinetic relationships for the system described are the same as those obtaining in the polymerization of acrylonitrile in other redox systems. From this the authors conclude that the primary stage of the polymerization proceeds not at the

Card 2/3

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Explosions. Topochemistry. Catalysis.

Abs Jour : Referat Zhur--Khimiya, No. 11, 1959, 37927

surface of the catalyst but in the solution.
For Communication XX see RZhKhim, 1959, 30579.
-- S. Rozenfel'd

Card 3/3

CSUROS, Z.; DEAK, GY.; VARSANYI, GY.

Examinations by catalysts. XXIX. Catalyzed anomerization of pentaacetylene-D-glucose with boron trifluoride. II. Anomerization in chloroform. p. 389.

Magyar Tudományos Akademia. Kémiai Tudományok Osztálya. KOZLEMENYEI. Budapest, Hungary, Vol. 10, No. 3, 1958.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, No. 7, July 1959
UNCL

Csuros, Z., and others.

Rheologic investigation of macromolecular substances. IV. Rheologic investigation of PVC pastes. II. Effect of temperature on PVC pastes . p. 467.

Magyar Tudományos Akadémia. Kémiai Tudományok Osztálya. KÖZLEMÉNYEI.
Budapest, Hungary, Vol. 10, No. 4, 1958

Monthly List of East European Accessions (EEAI) LC, Vol. 8, No. 7, July 1959

Uncl.